

# CHE 447: Heat and Mass Transfer in Chemical Engineering Processes Fall 2022

Note: All times are in Pacific Time; PST prior to March 13 and PDT after March 13

## Lectures

Mondays and Wednesdays, 2:00-3:50 PM, GFS 116 (After January 18)

## Instructor

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Office hours: Tuesdays and Thursdays 1:00-2:00 PM or by appointment

## Teaching Assistant

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Office Hours: via Zoom by appointment; usually before homework due dates and exams

## Course Description and Learning Objectives

This is an introductory course to mass and heat transfer in chemical engineering. Topics include molecular and continuum approaches to diffusion and convection in fluids. We will derive differential and macroscopic equations of mass and heat transfer and discuss their solutions for different geometries in different applications. The learning objectives are to understand the mass and heat transfer concepts and definitions, to construct the mathematical models, and to apply the modeling methods in chemical engineering designs.

First, we will study the advective mass transfer by reviewing the fluid flows in pipes and in porous media, including the mathematical models in terms of partial differential equations (PDEs), initial and boundary conditions. The commonly used solutions and applications will be presented. Next, we will study the diffusion phenomena and formulate the mathematical models (PDEs, initial and boundary conditions). Steady-state and transient solutions of the diffusion models will be derived and discussed for linear, radial, and spherical geometries. The applications of such solutions in chemical engineering will be explored. Afterward, we will study heat transfer methods, including convection, conduction, and radiation. Mathematical models of heat transfer will be formulated and solved for linear, radial, and spherical geometries. The applications of the heat transfer models in chemical engineering will be explored. Lastly, we will examine processes that utilize mass and heat transfer simultaneously.

Upon completion of this course, students should be able to identify, formulate, and solve complex engineering problems in the context of mass and heat transfer by applying the principles learned in this course.

## Books and References:

This course will be heavily based on the following textbook. Though you are NOT required to purchase this textbook, it is *highly recommended* that you do so as it will provide valuable supplementary reading material for this course.

- Stanley Middleman (1998) *An Introduction to Mass and Heat Transfer: Principles of Analysis and Design*, New York: John Wiley & Sons

## Lecture Schedule

This schedule is subject to change, with notice.

### Part 1: Mass Transfer

Mass transfer by advection  
Mass transfer in porous media, Darcy's law  
Diffusive Mass Transfer (Steady State)  
Convective Mass transfer (Steady State)  
Mass Transfer Coefficients  
Time-Dependent Mass Transfer

### Part 2: Heat Transfer

Heat transfer by conduction (Steady State)  
Heat transfer by convection (Steady State)  
Time-Dependent Heat Transfer  
Heat exchangers  
Natural convection  
Heat transfer by radiation  
Simultaneous mass and heat transfer

## Grading

Grades will be based on homework assignments, a midterm examination, a project, and a final examination.

|             |     |
|-------------|-----|
| Homework:   | 10% |
| Project:    | 10% |
| Midterm:    | 40% |
| Final Exam: | 40% |

Final letter grades will follow the scale below. However, a grading curve may be implemented at the instructor's discretion.

| <b>A</b> | <b>B+</b> | <b>B</b> | <b>C+</b> | <b>C</b>  | <b>D</b>  | <b>F</b> |
|----------|-----------|----------|-----------|-----------|-----------|----------|
| ≥ 90%    | [87%,90%) | [80,87%) | [77%,80%) | [70%,77%) | [60%,67%) | < 60%    |

## Homework

Homework will be assigned weekly on Wednesdays and due one week later the following Wednesday, unless there is an exam scheduled or the Wednesday falls on a university holiday. Homework assignments must be turned in as a photograph or scan on Blackboard. Homework assignments should be scanned/photographed such that they can be read easily (i.e., do not submit a low resolution .jpeg). Assignments will be graded according to a rubric that will be made available after the assignment due date. Late assignments will not be accepted unless there are genuine extenuating circumstances.

## Exams

Exams will be in-person with a time limit corresponding to the length of the class period. In the event that an in-person exam is not feasible, a take-home exam will be administered instead. Take-home exams will be due 24 hours after being distributed either physically

or as a scanned/photographed file via Blackboard. Exams will be graded according to a rubric that will be made available after the exams are graded. Exams cannot be rescheduled unless there are genuine extenuating circumstances.

### **Project**

Students will form groups (3-4 students per group) that critically address a theoretical (or real life) problem that is relevant to mass and/or heat transfer. These student groups will propose a scenario or process where the principles taught in this course are applicable to solving the problem. The project will be graded on originality, creativity, accuracy, and technical rigor. The projects will consist of both a written report (due April 22) and a group presentation (April 25 and additional presentations on April 27 if necessary). The detailed requirements for this group project will be provided after the midterm on March 9.

### **Special Dates**

Monday January 17: No class, Martin Luther King Jr. Day

Monday February 21: No class, President's Day

Wednesday March 9: Midterm Exam

Monday March 14: No Class, Spring Recess

Wednesday March 16: No Class, Spring Recess

Friday April 22: Written Project Reports Due

Monday April 25: Project Presentations

Wednesday April 27: Project Presentations (cont.), Last Day of Class

Monday May 9: Final exam (2:00-4:00 PM)

### **Collaboration Policy**

Students are encouraged to discuss and work together on homework assignments, but the work each student hands in must be their own. It is not acceptable to merely copy another student's effort; each student must be capable of fully understanding and describing everything they have written in the submitted homework assignment. To ensure that this is the case, it is recommended that when working in a group (group sizes of five or fewer are recommended), students plan their approach to a problem making notes on scratch paper or a blackboard. The work that each student submits, however, should be written independently without referencing these notes. If you have any doubts regarding whether a certain instance of collaboration is acceptable at any point in the semester, ask the instructor for clarification.

Work on exams will be completely independent.

### **Software**

All commercial computer software used in this course must be properly licensed. Use of unlicensed commercial software is not allowed and it may result in a failing grade. MATLAB and Microsoft Office licenses are available for free to students through USC.

### **Academic Conduct**

Plagiarism—presenting someone else's ideas as your own, either verbatim or recast in your own words—is a serious academic offense with serious consequences. Please familiarize yourself with the discussion of plagiarism in SCampus in Section 11, Behavior Violating University Standards:

<https://scampus.usc.edu/1100-behavior-violating-university-standards-and-appropriate-sanctions/>

Other forms of academic dishonesty are equally unacceptable. See additional information and university policies on scientific misconduct:

<http://policy.usc.edu/scientific-misconduct>

Discrimination, sexual assault, and harassment are not tolerated by the university. You are encouraged to report any incidents to the Office of Equality and Diversity (<http://equity.usc.edu>) or to the Department of Public Safety:

<http://adminopsnet.usc.edu/department/departement-public-safety/online-forms/contact-us>

This is important for the safety of the whole USC community. Another member of the university community—such as a friend, classmate, advisor, or faculty member—can help initiate the report, or can initiate the report on behalf of another person. The Center for Women and Men:

<http://www.usc.edu/student-affairs/cwm>

provides 24/7 confidential support, and the sexual assault resource center webpage (<https://sarc.usc.edu>) describes reporting options and other resources.

### **Support Systems**

A number of USC's schools provide support for students who need help with scholarly writing. Check with your advisor or program staff to find out more. Students whose primary language is not English should check with the American Language Institute (<http://dornsife.usc.edu/ali>), which sponsors courses and workshops specifically for international graduate students. The Office of Disability Services and Programs:

[http://sait.usc.edu/academicsupport/centerprograms/dsp/home\\_index.html](http://sait.usc.edu/academicsupport/centerprograms/dsp/home_index.html)

provides certification for students with disabilities and helps arrange the relevant accommodations. If an officially declared emergency makes travel to campus infeasible, USC Emergency Information (<http://emergency.usc.edu>) will provide safety and other updates, including ways in which instruction will be continued by means of Blackboard, teleconferencing, and other technology.