

# AME 513

## Principles of Combustion

Fall 2017, OHE 100C, F 9:00-11:50 am

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### References:

1. Combustion Physics, by C.K. Law, 1<sup>st</sup> Edition, Cambridge University Press, 2006, (required).
2. Unpublished notes, prepared by C.K. Law & F.N. Egolfopoulos (will be provided).
3. Combustion Theory, by Forman A Williams, 2<sup>nd</sup> Edition, Addison-Wesley, 1985.
4. Combustion, Flames, and Explosions of Gases, by Bernard Lewis and Guenther von Elbe, 3<sup>rd</sup> Edition, Academic Press, 1987.
5. Combustion, by Irvin Glassman, 3<sup>rd</sup> Edition, Academic Press, 1996.
6. An Introduction to Combustion, Concepts and Applications, by Stephen R. Turns, 2<sup>nd</sup> Edition, McGraw-Hill, 2000.
7. Molecular Theory of Gases and Liquids, by Joseph O. Hirschfelder, Charles F. Curtiss, and R. Byron Bird, 2<sup>nd</sup> Edition, John Wiley & Sons, 1963.
8. Physical Chemistry, by P.W. Atkins, 4<sup>th</sup> Edition, W.H. Freeman and Company, New York, 1990.
9. Chemical Kinetics, by Keith Laidler, 3<sup>rd</sup> Edition, Harper and Row, 1987.
10. Thermochemical Kinetics, by Sidney W. Benson, John Wiley & Sons, 1968.

### Topics:

Introduction  
Chemical Thermodynamics  
Chemical Kinetics  
Transport Phenomena  
Conservation Equations  
Non-Premixed Flames  
Premixed Flames  
Aerodynamics of Laminar Flames  
Environmental Combustion Considerations

<b>Grading:</b>	Midterm Exam	TBD	35%
	Final Exam	December 11 (M) (11:00 am-1:00 pm)	45%
	Homework		20%

**Note:** The use of laptops or cell phones to access the internet/e-mail during class and/or exams is not allowed. Such devices are allowed only to access material pertaining to the class.