AME 513

Principles of Combustion Fall 2013, OHE 100C, F 9:00-11:50 am

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	Office Hours:	Anytime by appointment		

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References:

- 1. Combustion Physics, by C.K. Law, 1st Edition, Cambridge University Press, 2006, (required).
- 2. Unpublished notes updated yearly, by C.K. Law & F.N. Egolfopoulos (will be provided).
- 3. Combustion Theory, by Forman A Williams, 2nd Edition, Addison-Wesley, 1985.
- 4. Combustion, Flames, and Explosions of Gases, by Bernard Lewis and Guenther von Elbe, 3rd Edition, Academic Press, 1987.
- 5. Combustion, by Irvin Glassman, 3rd Edition, Academic Press, 1996.
- 6. An Introduction to Combustion, Concepts and Applications, by Stephen R. Turns, 2nd Edition, McGraw-Hill, 2000.
- Molecular Theory of Gases and Liquids, by Joseph O. Hirschfelder, Charles F. Curtiss, and R. Byron Bird, 2nd Edition, John Wiley & Sons, 1963.
- 8. Physical Chemistry, by P.W. Atkins, 4th Edition, W.H. Freeman and Company, New York, 1990.
- 9. Chemical Kinetics, by Keith Laidler, 3rd Edition, Harper and Row, 1987.
- 10. Chemical Kinetics of Gas Reactions, by V.N. Kondrat'ev, Pergamon Press, 1964.
- 11. Physical Chemistry of Fast Reactions, Volume 1, Gas Phase Reactions of Small Molecules, edited by B.P. Levitt, Plenum Press, 1973.
- 12. Thermochemical Kinetics, by Sidney W. Benson, John Wiley & Sons, 1968.

Topics:

Introduction Chemical Thermodynamics Chemical Kinetics Transport Phenomena Conservation Equations Non-Premixed Flames Premixed Flames Aerodynamics of Laminar Flames Environmental Combustion Considerations

Grading:	Midterm Exam	October 18 (F)	(9:00 am-10:30 am)	35%
	Final Exam	December 9 (M)	(11:00 am-1:00 pm)	45%
	Homework			20%

Note: The use of laptops or cell phones to access the internet/e-mail during class and/or exams is not allowed.